SYNTHESIS OF THE CYCLIC TETRAPEPTIDE TENTOXIN. EFFECT OF AN N-METHYLDEHYDROPHENYLALANYL RESIDUE ON CONFORMATION OF LINEAR TETRAPEPTIDES

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School of Pharmacy, University of Wisconsin, Madison, Wisconsin 53706 (Received in USA 8 August 1974; received in UK for publication 8 October 1974) Tentoxin, a cyclic tetrapeptide produced by the phytopathogenic fungus, <u>Alternaria tenuis</u>, causes chlorosis when applied to germinating seed.¹ Recent studies have proposed that tentoxin is cyclo(<u>N-methyl-L-alanyl-L-leucyl-N-methyl-Z-dehydrophenylalanyl-glycyl</u>) (<u>8</u>).² We report here the first synthesis of the cyclic tetrapeptide <u>8</u>, which confirms the proposed structure, and present n.m.r. and c.d. evidence which suggests that the conformation of the linear tetrapeptide <u>5a</u> is similar to cyclic peptide 8.

<u>N-t</u>-Butoxycarbonyl-<u>N</u>-methyl-L-alanyl-L-leucyl-3-benzylthio-D,L-phenylalanyl-glycyl-<u>O</u>-resin (<u>1</u>) was prepared by solid phase synthesis from <u>N-t</u>-butoxycarbonyl-glycyl-<u>O</u>-resin (0.6 mmol Gly/g, 1% divinyl benzene) according to the general procedure of Merrifield. ³ <u>N-t</u>-Butoxycarbonyl-3benzylthio-D,L-phenylalanine⁵ was prepared from 3-benzylthio-D,L-phenylalanine⁴ and used as the mixture of diastereomers. Methanolysis of the resin <u>1</u> gave the diastereomeric sulfides <u>2a,b</u>⁵ in an overall yield of 64%. Treatment of sulfides <u>2a,b</u> with sodium periodate in aqueous methanol for 24 h gave the sulfoxides <u>3a,b</u> (yield 94%)⁵ which were converted to the dehydrophenylalanyl peptides <u>4a,b</u> (60% yield)⁵ by heating in refluxing xylene under an atmosphere of nitrogen for 30 h.⁶ The geometric isomers <u>4a</u> (R_f, EtOAc = 0.62) and <u>4b</u> (R_f, EtOAc = 0.81) were separated by column chromatography on silica gel. Peptide <u>4a</u> (λ_{max} 276, £18,400) was assigned the <u>Z</u>-configuration and peptide <u>4b</u> (λ_{max} 282, £9080) assigned the <u>E</u>-configuration on the basis of the greater intensity of absorbance observed for the <u>trans</u>-isomer of cinnamic acids.⁷ These assignments are consistent with the observations that the chemical shifts (Table) of the leucyl and alanyl protons in <u>4a</u> (Z) are shielded with respect to the same protons in <u>4b</u> (E) while the glycyl protons are shielded in <u>4b</u>.

Peptide <u>4a</u> was transformed to the <u>N</u>-methyl-<u>Z</u>-dehydrophenylalanyl peptide <u>5a</u> (89% yield)⁵ by reaction with methyl iodide and potassium carbonate in dimethylformamide at room temperature for 72 h. This methylation reaction was found to proceed stereospecifically on model compounds. For example, the methylesters of Z- and E- N-t-butoxycarbonyl-3-methyldehydroalanine were converted to the Z- and E- N-methyl derivatives respectively in high yield without isomerization of the double bond.⁸ No methylated products (either the E- or Z- isomers) were detected when the E-dehydrophenylalanyl isomer <u>4b</u> was subjected to the same methylation procedure. Peptide <u>4b</u> was recovered in high yield.

The conformation of peptide 5a (N-CH₃- Δ Phe) is substantially different from that of peptide 4a (NH- Δ Phe). The chemical shifts (Table) of the α - and δ -protons of leucine, and the methyl and vinyl protons of N-methyl-Z-dehydrophenylalanine in 5a are nearly the same as those determined for tentoxin, <u>8</u>. The leucine resonances in both noncyclic precursor 5a and natural cyclic peptide <u>8</u> are shifted upfield from what one finds in other linear dehydropeptides (<u>4a,b;</u> <u>9</u>) which contain the dehydrophenylalanyl unit suggesting that the orientation of the leucyl isobutyl group with respect to the phenyl ring in both 5a and 8 is similar. In contrast, peptide 4a which contains three secondary amide bonds should be in an all trans amide bond conformation similar to 10a. The chemical shift of the leucyl isobutyl group in 10a should be normal (about 0.91) since the isobutyl group is too far from the phenyl ring to be shielded by it. The circular dichroism spectra of 5a (277 nm, $[\Theta]$ + 6640) and 8 (285 nm, $[\Theta]$ 3593) showed comparable positive elipticities in the 260-320 nm region, indicating similar stereochemical and electronic environments about the α,β -dehydro chromophore. Unmethylated peptide <u>4a</u> showed a strong negative elipticity in the same region (275 nm, [0]-12,212). The close similarity between the n.m.r. and c.d. spectra of 5a and 8 suggests that the preferred conformation of 5a resembles structure 11 (and not 10b) in that the amide bond between the N-methyldehydrophenylalanyl and leucyl residues is predominently the cis conformer.

The protected peptide acid $(\underline{6}, R = H)$,⁵ obtained from ester <u>5a</u> by saponification (93% yield), was converted to the 2,4,5-trichlorophenyl ester (<u>7</u>. $R = C_6H_2Cl_3$) by reaction with dicyclohexylcarbodiimide in pyridine. After removal of the <u>N</u>-t-butoxycarbonyl group by treatment of <u>7</u> with 1 <u>N</u> hydrochloric acid in acetic acid, the resulting peptide was cyclized by reaction for 24 h in refluxing pyridine (0.18 mmolar) containing one equivalent of diisopropylethylamine. The cyclic tetrapeptide <u>8</u> was purified by chromatography and isolated in a yield of 18% based on the protected acid <u>6</u>. The n.m.r., i.r., u.v. mass spectral data and the chromatographic behavior of synthetic <u>8</u> were identical with those of natural tentoxin. The biological potencies of synthetic and natural tentoxin on germinating lettuge seedlings were indistinguishable. These data confirm the sequence reported for tentoxin.² This work was supported by grants from the National Institutes of General Medical Studies, the Petroleum Research Foundation, and the Graduate School of the University of Wisconsin. We thank Professor R. D. Durbin for biological data, and Professor P. A. Hart for c.d. data.



Compound	Leu		<u>Ala</u>			Gly		Me Phe	
	δĦ	<u>а, Н</u>	β-Н	N-CH3	<u>a-H</u>	<u>α-H</u>	CO2CH3	N-CH3	<u>β−Н</u>
<u>4a</u>	0.91	4.4	1.27	2.77	4.50	4.05 (d, J=6)	3.71		
<u>4b</u>	0.92	4.56	1.36	2.79	4.50	3.91 (d, J=6)	3.64		
<u>5a</u>	0.54 0.60	4.18	1.26	2.70	4.66	4.14 (d, J=6)	3.71	3.21	7.74
<u>8</u>	0.53 0.63	4.18	1.53	2.80	4.37	5.21 (d,d; J=10,9) 3.57 (d, J=15)		3.18	7.75
<u>9</u>	0.93	4.6		2.96		4.16 4.22	3.71	3.05	7.35

++The spectra were obtained with a Bruker-HX90E spectrometer in deuteriochloroform. The chemical shifts are expressed in ppm relative to internal TMS.

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